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Aromatic Heterocycles as Activating Groups for Asymmetric Conjugate Addition Reactions. Enantioselective Copper-Catalyzed Reduction of 2-Alkenylheteroarenes

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Table 1. Ligand Optimization for the Asymmetric Reduction of 1a^a

The 1,4-addition of a nucleophile to an alkene conjugated to an electronwithdrawing group is a fundamental reaction in organic chemistry, and numerous catalytic asymmetric variants (eq 1) of this process are now routinely employed in the synthesis of molecules of interest.^{1,2} The most common functional groups used to activate alkenes toward asymmetric conjugate additions include carbonyls, nitriles, sulfones, phosphonates, and nitro groups. We recently questioned whether other rarely considered yet common functional groups might also be employed in this capacity, and our attention focused on nitrogen-containing aromatic heterocycles. Given that heteroarenes such as oxazoles, thiazoles, pyridines, and others are ubiquitous in biologically active natural products, pharmaceuticals, and agrochemicals, the ability to functionalize these privileged structures through a diverse set of asymmetric conjugate additions of 2-alkenyl derivatives (eq 2) would open up broad-ranging applications.



Although conjugate additions to 2-vinylheteroarenes (R¹, R² = H in eq 2) are relatively common,³ the corresponding reactions of substrates containing a β -substituent are much rarer, presumably for steric reasons.^{4,5} Furthermore, the only report of a catalytic enantioselective variant is limited to poorly selective ($\leq 15\%$ ee) Grignard additions to 4-alkenylpyridines.⁵ Therefore, we recently initiated a program targeted at addressing these deficiencies, and in this communication, our preliminary findings involving highly enantioselective copper-catalyzed reductions⁶ of β , β ³-disubstituted 2-alkenylheteroarenes are presented.

The asymmetric copper-catalyzed conjugate reduction of activated alkenes is a well-established method for the synthesis of various useful chiral building blocks.^{6–10} Whether a nitrogen-containing heteroarene would provide sufficient activation to an adjacent alkene in an analogous reaction was, however, uncertain. In addition, it seemed likely that coordination of the Lewis basic nitrogen of the heteroarene to the copper catalyst would occur in such a process, and whether this interaction would be beneficial, inconsequential, or detrimental was not easy to predict.

Our investigations began with a survey of chiral bisphosphines L1–L6 using 2-alkenylbenzoxazole 1a as a test substrate (Table 1). Using 10 mol % of Cu(OAc)₂·H₂O, 10 mol % of ligand, and 4 equiv each of PhSiH₃ and *t*-BuOH in toluene at room temperature, biaryl-based ligands L1–L4 proved competent in promoting conjugate reduction.¹¹ With (*R*)-BINAP (L1), both conversion and enantioselectivity were only moderate. However, improved results were observed using (*R*)-MeO-BIPHEP (L2) and the



^{*a*} Reactions were conducted using 0.20 mmol of **1a** in toluene (1 mL). Conversions were determined by GC analysis. Enantioselectivities were determined by chiral HPLC analysis. ^{*b*} Reactions complete after 2 h.

SEGPHOS ligands L3 and L4, with 91% ee obtained using (*S*)-SEGPHOS (L3). The Josiphos ligands L5 and L6 were also effective, providing 2a in 89% and 87% ee, respectively. Of all the ligands, the highest reaction rates were observed with L4 and L5 (reactions were complete in 2 h). However, the superior selectivity provided by L5 prompted us to select this ligand for further optimization and investigation of the reaction scope.

Using 5 mol % each of Cu(OAc)₂·H₂O and **L5**, PhSiH₃ (1.5 equiv), and *t*-BuOH (2.0 equiv) at an initial temperature of 0 °C, a range of β , β' disubstituted 2-alkenylheteroarenes underwent conjugate reduction with generally excellent levels of enantioselection (Table 2).¹¹ In addition to benzoxazole (entries 1–4), other effective nitrogen-containing heteroarenes in this process included 5-phenyloxazole (entry 5), benzothiazole (entry 6), pyridine (entries 7–9), quinoline (entry 10), and pyrazine (entry 11). Tolerated functionality at the β -positions of the alkene included simple aliphatic groups, a phenyl group (entry 4), a benzyl group (entry 9), various oxygenated alkyl groups (entries 2, 3, 7–9, and 11), and a cyclopropane (entry 10). The process is tolerant of lower loadings of copper and ligand. For example, reduction of **1g** on a 1.0 mmol scale using 2 mol % each of Cu(OAc)₂·H₂O and **L5** provided **2g** in 92% yield and 96% ee (entry 7, values in parentheses).

Experiments to explore the origins of reactivity were then conducted. Reduction of 4-alkenylpyridine **3** provided **4** in 60% yield and 94% ee, albeit in a slower reaction that was incomplete even after 4 days (eq 3). This result suggests that alkene reduction by copper hydride can occur without assistance of a directing effect from the nitrogen atom. In contrast,
 Table 2.
 Scope of Cu-Catalyzed Asymmetric Conjugate

 Reduction^a
 Provide Asymmetric Conjugate



^{*a*} Reactions were conducted using 0.20 mmol of **1a–1k** unless otherwise stated. ^{*b*} Isolated yield. ^{*c*} Determined by chiral HPLC analysis. ^{*d*} Using 0.155 mmol of **1d**. ^{*e*} Values in parentheses refer to a reaction conducted using 1.0 mmol of **1g**, 2 mol % Cu, and 2 mol % **L5**. ^{*f*} Using 0.10 mmol of **1h** and 2.0 equiv of PhSiH₃.

3-alkenylpyridine **5** was unreactive (eq 4), demonstrating the importance of conjugation of the alkene to a C=N moiety for reactivity.



In summary, copper-catalyzed asymmetric conjugate reductions of β , β' disubstituted 2-alkenylheteroarenes are reported. In addition to serving as a further demonstration of the power of chiral copper hydride catalysis,⁶ this work has shown nitrogen-containing aromatic heterocycles can provide effective activation of an adjacent alkene for highly enantioselective catalytic conjugate addition reactions. Extension of the general concept to other classes of asymmetric reactions should provide a range of useful tools for chemists working with heteroarenes. Future studies from our laboratory will be directed toward this goal.

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Supporting Information Available: Experimental procedures, full spectroscopic data for new compounds, and crystallographic data in cif format. This material is available free of charge via the Internet at http://pubs.acs.org.

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